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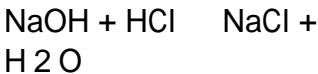
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Answers

Solution: The
reaction of the acid +
base gives a product
of salt + water, which
is considered as

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reaction. Examples:



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Question 1. Two

solutions A and B

have pH values of 5
and 8 respectively.

Which solution will

be basic in nature ...

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Ans:- Salts are classified into four categories on the basis of the strength of acid and base. they are: Salt of strong acid and strong base.e.g NaCl, KCl; Salt of strong acid and weak base. e.g. NH_4Cl , NH_4I ; Salt of

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Salts of strong acid and strong base. e.g. CH_3COONa , $\text{COO}^- \text{Na}^+$ +
Salt of weak acid and weak base. e.g. $\text{CH}_3\text{COO}^- \text{NH}_4^+$ + How is common salt found in nature?

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2. Bases react with .
acids to produce salts
and water; water to
produce acids and
salts; salts to produce
acids and water;
neither acids, salts,
nor water

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The salts of strong acids and weak bases give acidic solution having pH less than 7. Example, NH_4Cl , Ammonium Chloride will have pH less than 7. The salts of weak acids and strong

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bases give basic
solution having pH
more than 7.

Example, Na_2CO_3 ,
Sodium Carbonate
will have pH more
than 7. Question 7.

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Answer A reaction in
which an acid and
base react with each
other to give a salt
and water is termed
as neutralization

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reaction. For Example:



15. Give two

important uses of

washing soda and

baking soda. Answer

Two important uses

of washing soda are:

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Answer. (a) HCl is acid and NaOH is base whose combination forms the common salt. Its formula is NaCl (Sodium chloride). It is obtained from sea water. (b) Rock salt is the common name for the mineral " halite " . Its chemical formula is NaCl. It may be white

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or light blue or yellow
depending upon
impurities present in
it.

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(A) inorganic acid (B)
organic acid (C) soft
organic acid (D)
strong inorganic acid.

Answer. Answer: (a)

5. When milk of
magnesia reacts with
acetic acid it
produces (A) basic
salts (B) acidic salts
(C) neutral salt (D)

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complex salt. Answer.

Answer: (c) 6. Which
of the following
phenomena will
occur when a small
amount of acid is
added to water ...

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Turmeric is a natural indicator. On adding its paste to acid and base separately, which colours would be observed. (a) Yellow in both acid and base (b) Yellow in acid and red in base (c) Pink in acid and yellow in base (d) Red in acid and blue in base. Answer.

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Answer: (b) Yellow in
acid and red in base

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alkalis react with one another, how neutralisation can produce salts and the relevance of pH values.

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An acid is a
compound that is
defined by its

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physical and
chemical properties.
Acids taste sour and
react with metals and
polyatomic ions
called carbonates. A
carbonate is a
charged cluster of
Carbon and Oxygen
atoms.

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Acid + Base ---> Salt

+ Water. Question 3:

Rain becomes acidic

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due to. carbon dioxide. sulphur dioxide. nitrogen dioxide. all of the above. Correct

Option is : 4. Solution : Rain becomes acidic because of carbon dioxide, sulphur dioxide and nitrogen dioxide. They dissolve in rain drops to form carbonic acid, sulphuric acid and

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