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change in chemical properties of a substance like composition, reactivity with water, corrosiveness, and pH is known as chemical change. In chemical change, a new substance will be formed.

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are mixed, forming the precipitate silver iodide. chemical equation: KI(aq) AgNO 3(aq) 0 KNO 3(aq) AgI(s) complete ionic equation: K(aq) I (aq) 32 Ag (aq) NO 3 (aq) 0 K (aq) NO 3 (aq) AgI(s)

Chemical ReactionsChemical Page 24/38

Download Ebook Chemistry **Matter And Change** Reactions Manual Chapter Solution. Physical: boiling and melting are physical changes. When water boils, no bonds are broken or formed. The change could be written: \(\ce{H\_2O}\) \left( | \right) \rightarrow \ce{H\_2O} \left( g \right)\) Page 25/38

Chemical: The dark grey nail ter changes color to form an orange flaky substance (the rust); this must be a chemical change.

3.6: Changes in Matter: Physical and Chemical Changes ...240 Chemistry: Matter and Change Page 26/38

 Chapter 12 Solutions Manual CHAPTER 12 SOLUTIONS MANUAL 30. Compare and contrast sublimation and evaporation. In both processes, the substances become a vapor. During sublimation, the substance goes from the solid phase directly Page 27/38

to the vapor phase. During apter evaporation, particles in a liquid gain enough

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