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Physics Glencoe Physics: Principles
and Problems, Student Edition The de
Broglie wavelength and speed of an
electron. The De-Broglie wavelength is
 $= 7.77 \times 10^{-11} \text{ m}$ The speed of an
electron is $= 9.376 \times 10^6 \text{ m s}^{-1}$
Given data: The electron accelerated

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with a potential difference (V) = 250

V Formula used: $KE = \frac{1}{2} m v^2 = e V$

Where, KE = kinetic energy m = mass

e = charge of electron v = velocity V =

potential difference Re-arrange the

equation, $v = \left(\frac{2 e V}{m} \right)^{1/2}$ is the

required ...

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