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For methods where bias is applicable (see Table 2020:1), run a laboratory-fortified blank (LFB) (2020B.2e), performance evaluation sample, or standard with a known or otherwise certifiable concentration at least four times and compare results to the limits listed in the method or those established by the laboratory. If no limit is specified, use the following procedure to establish limits:

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where:  $\text{pH} \uparrow = \log_{10} a_{\text{H}^+}$  and  $\text{pOH} = \log_{10} a_{\text{OH}^-}$ . Equation 2 states that as pH increases pOH decreases correspondingly and vice versa because  $\text{pK}_w$  is constant for a given temperature. At 25 ° C, pH 7.0 is neutral, the activities of the hydrogen and hydroxyl ions are equal, and each corresponds to an approximate activity of  $10^{-7}$  moles/L. The neutral point is temperature-dependent and is ...

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